

TRITERPENOIDS FROM *MYRICA RUBRA**

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Key Word Index *Myrica rubra*; Myricaceae; taraxerone; taraxerol; myricadiol; 28-hydroxy-D-friedoolean-14-en-3-one.

Abstract—A new triterpene has been isolated together with taraxerone, taraxerol, myricadiol and sitosterol from the stem bark of *Myrica rubra*. On the basis of chemical and spectral evidence, the structure was established as 28-hydroxy-D-friedoolean-14-en-3-one.

INTRODUCTION

The bark of *Myrica rubra* Sieb. et Zucc. has been used in Japan and China as an astringent, an antidote and as an antidiarrhoea agent.

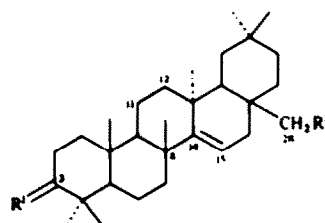
Recently, diarylheptanoid [1], a galloyl flavanonol sulphate, tannin and related compounds [2] were isolated from the stem bark of *M. rubra*. We have examined the same source and report the isolation and characterization of a new triterpenoid together with taraxerone, taraxerol, myricadiol and sitosterol.

RESULTS AND DISCUSSION

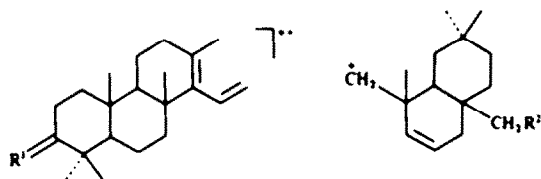
The benzene extract of the stem bark was separated into acidic and neutral fractions, and compounds 1–5 were isolated from the neutral fraction. Compound 1, $C_{30}H_{48}O$, mp 240–243°, gave a mass spectrum which showed intense peaks at m/z 300 (ion a) and 204 (ion b) (Scheme 1). Compound 1 was identified as taraxerone by comparison with an authentic sample [3]. Compound 2, $C_{30}H_{50}O$, mp 278–279°, gave a mass spectrum which showed intense peaks at m/z 302 (ion c) and 204 (ion b). Compound 2 was identical with an authentic sample of taraxerol [4]. Compound 3, mp 137–140°, was identified as sitosterol by its spectral data.

Compound 4, $C_{30}H_{50}O_2$, mp 271–272°, gave a mass spectrum which showed intense peaks at m/z 302 (ion c) and 220 (ion d). Acetylation of 4 with acetic anhydride in pyridine afforded a diacetate (4a), $C_{34}H_{54}O_4$, mp 259–260°, and a monoacetate (4b), $C_{32}H_{52}O_3$, mp 249–250°. The IR and 1H NMR spectral data of 4a were in good agreement with those of an authentic sample of myricadiol diacetate [4] and thus compound 4 was identified as myricadiol.

Compound 5, mp 225–227°, $[\alpha]_D -0.2^\circ$, analysed for $C_{30}H_{48}O_2$ and its IR spectrum showed hydroxyl (3350 , 1000 cm^{-1}) and carbonyl (1708 cm^{-1}) absorption. Its 1H NMR spectrum exhibited a double doublet of an olefinic proton at $\delta 5.54$ ($J = 8.1$ and 3.4 Hz) and two AB-type doublets at $\delta 3.14$ and 3.30 ($J = 10.0$ Hz) due to a



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|----|--|
| 1 | $R^1 = O, R^2 = H$ |
| 2 | $R^1 = \alpha-H, \beta-OH, R^2 = H$ |
| 4 | $R^1 = \alpha-H, \beta-OH, R^2 = OH$ |
| 4a | $R^1 = \alpha-H, \beta-OAc, R^2 = OAc$ |
| 4b | $R^1 = \alpha-H, \beta-OH, R^2 = OAc$ |
| 5 | $R^1 = O, R^2 = OH$ |
| 5a | $R^1 = O, R^2 = OAc$ |



- | | |
|--|------------------------------|
| ion a m/z 300, $R^1 = O$ | ion b m/z 204, $R^2 = H$ |
| ion c m/z 302, $R^1 = \alpha-H, \beta-OH$ | ion d m/z 220, $R^2 = OH$ |
| ion e m/z 344, $R^1 = \alpha-H, \beta-OAc$ | ion f m/z 262, $R^2 = OAc$ |

Scheme 1. Mass spectral fragments derived from compounds 1, 2, 4 and 5.

hydroxymethylene group. In the mass spectrum of 5, strong peaks at m/z 300 (ion a), 220 (ion d) and 189 (ion d – CH_2OH) indicated the presence of a double bond at C-14 and the hydroxymethylene group at C-28 having no substituents in rings C, D and E. Acetylation of 5 afforded a monoacetate (5a), $C_{32}H_{50}O_3$, mp 210–211°. From the above result, we can deduce that compound 5 is 28-hydroxy-D-friedoolean-14-ene with a carbonyl group probably at C-3.

*Part 2 in the series "Constituents of *Myrica rubra*". For Part 1, see ref. [1].

The skeletal structure was confirmed by chemically relating myricadiol (4) to compound 5. Oxidation of myricadiol monoacetate (4b) with chromium trioxide pyridine gave the monoacetate (5a). Deacetylation of 5a afforded 28-hydroxy-D-friedoolean-14-en-3-one, which was identical with compound 5.

The ^{13}C NMR chemical shifts of compounds 1–5 were assigned by comparison with the reported shifts of aleuritonic acid (6) and acetoxyaleuritolate (7) [5], and are shown in Table 1. In the ^{13}C NMR spectrum of 5, all signals were in good agreement with the proposed structure.

Taraxerone, taraxerol and myricadiol were isolated from the root bark of *M. cerifera* [3], taraxerol and myricadiol from the bark of *M. esculenta* [4], and myricadiol from the stem bark of *M. gale* [6] and *M. nagi* [7]. Taraxerol was also found in the leaves of *M. rubra* [8]. From the chemotaxonomic view, these triterpenoids seem to be widely distributed in *Myrica* species.

EXPERIMENTAL

All mps were uncorr. ORDs were measured using a 1 dm cell. ^1H NMR spectra were taken at 100 MHz in CDCl_3 soln using

TMS as internal standard. ^{13}C NMR spectra were recorded at 25 MHz in CDCl_3 (TMS as internal standard). MS were run on a double focusing mass spectrometer (accelerating voltage of 3.0–6.5 kV; ionizing potential 70 eV). TLC was carried out on silica gel.

Extraction and isolation of constituents. Air-dried and ground stem bark of *M. rubra* (4 kg) was extracted with C_6H_6 . The C_6H_6 soln was coned *in vacuo* to 2 l, and 2 M NaOH was added to the extract. The NaOH soln was partitioned with EtOAc. The EtOAc extract (20.0 g) was chromatographed repeatedly on silica gel using solvent systems of C_6H_6 -EtOAc and CHCl_3 -MeOH to give 1 (560 mg), 2 (2.7 g), 3 (530 mg), 4 (2.7 g) and 5 (26 mg).

Taraxerone (1). Colourless plates, mp 240–243° (CHCl_3 -MeOH) (lit. [3] 238–239°), $[\alpha]_D^{25} + 9.6^\circ$ (CHCl_3 ; c 1.01). Taraxerone (1) was identified by direct comparison with an authentic sample (mmp, TLC, ^1H NMR). ^{13}C NMR: see Table 1.

Taraxerol (2). Colourless needles, mp 278–279° (C_6H_6) (lit. [4] 273–274°), $[\alpha]_D^{25} + 0.5^\circ$ (CHCl_3 ; c 1.10). Taraxerol (2) was identified by direct comparison with an authentic sample (mmp, TLC, ^1H NMR). ^{13}C NMR: see Table 1.

Sitosterol (3). Colourless needles, mp 137–140° (EtOH) (lit. [9] 138–139°). Sitosterol (3) was identified by direct comparison with an authentic sample (TLC, IR).

Myricadiol (4). Colourless needles, mp 271–272° (CHCl_3 -MeOH) (lit. [4] 268–269°). ^{13}C NMR: see Table 1.

Table 1. ^{13}C NMR chemical shifts of taraxerone (1), taraxerol (2), myricadiol (4), compound 5, the acetate of 5 (5a), aleuritonic acid (6) and acetoxyaleuritolate (7)

Carbon No.	1	2	4	5	5a	6	7
C-1	38.4	38.1	37.8	38.3	38.4	38.4	37.4
C-2	34.1	27.3	28.0	34.0	34.1	34.1	23.4
C-3	217.3	79.2	78.2	217.2	217.2	217.3	80.8
C-4	47.6	39.1	41.4	47.5	47.6	47.5	37.6
C-5	55.8	55.7	56.0	55.7	55.8	55.7	55.6
C-6	20.0	19.0	19.2	19.9	19.9	21.5	18.7
C-7	35.2	35.3	36.3	35.8	35.7	35.4	35.3
C-8	38.9	38.9	39.3	39.0	39.0	38.9	39.0
C-9	48.7	48.9	45.6	48.6	48.6	48.6	49.0
C-10	37.6	37.9	37.8	37.5	37.4	37.3	37.3
C-11	17.5	17.7	17.9	17.3	17.3	17.3	17.3
C-12	35.8*	35.9*	31.2*	30.7*	31.0*	31.3*	31.2
C-13	37.7	37.9	38.3	37.7	37.7	37.8	37.9
C-14	157.6	158.1	158.7	158.5	158.3	162.1	160.5
C-15	117.2	117.0	116.8	116.0	116.2	117.1	116.8
C-16	36.7*	36.9*	33.2*	32.6*	32.6*	30.7	30.6
C-17	37.7	38.1	38.3	40.3	39.0	51.5	51.5
C-18	48.8	49.4	49.6	44.9	44.6	41.4	41.3
C-19	40.7	41.4	41.7	40.6	40.6	40.3	40.7
C-20	28.8	29.0	28.8	28.5	28.5	29.3	29.3
C-21	33.6	33.9	33.8	33.3	33.5	33.7	33.6
C-22	33.1	33.2	28.7	27.9	28.3	31.9*	31.8
C-23	26.2	28.1	28.4	26.1	26.1	26.1	27.9
C-24	21.5	15.6	16.5	21.6	21.8	20.0	16.6
C-25	14.8	15.6	15.7	14.8	14.8	15.0	15.7
C-26	29.9	30.1	30.1	29.8	29.8	28.7	28.6
C-27	25.6	26.0	26.2	25.7	25.7	25.9	26.2
C-28	29.9	30.1	64.6	65.4	65.9	184.7	184.4
C-29	33.4	33.5	33.8	33.5	33.3	33.2	33.3
C-30	21.4	21.5	22.0	21.4	21.5	22.6	22.4

*Values in any vertical column may be reversed although those given here are preferred.

Acetylation of 4. A soln of **4** (200 mg) in pyridine (50 ml) was acetylated with Ac_2O (0.09 ml) at room temp. for 30 hr. After usual work-up, the crude product was chromatographed on silica gel to give the diacetate (**4a**, 10 mg) and the monoacetate (**4b**, 98 mg). Diacetate of **4** (**4a**), colourless needles, mp 259–260° (EtOH) (lit. [4] 252–253°), $[\alpha]_D^{25} + 6.9^\circ$ (CHCl_3 ; c 1.00). The diacetate of **4** (**4a**) was identified by direct comparison with an authentic sample (mmp, TLC, IR). Monoacetate of **4** (**4b**), colourless needles, mp 249–250° (CHCl_3 -EtOH), $[\alpha]_D^{25} + 3.0^\circ$ (CHCl_3 ; c 1.01). (Found: C, 79.28; H, 10.93. $\text{C}_{32}\text{H}_{52}\text{O}_3$ requires: C, 79.28; H, 10.81%). IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} : 3620, 1020 (OH), 1730, 1220 (OAc). $^1\text{H NMR}$: δ 0.80, 0.90, 0.92 (3H each), 0.96 (6H), 0.97, 1.08 (3H each) (all s, Me \times 7), 2.03 (3H, s, OAc), 3.18 (1H, m, H-3), 3.72 (2H, s, H₂-28), 5.45 (1H, dd, $J = 8.2, 3.7$ Hz, H-15). MS m/z : 484 $[\text{M}]^+$, 466 $[\text{M} - \text{H}_2\text{O}]^+$, 302 (ion c), 262 (ion f), 202 (ion f - HOAc), 189 (base peak) (ion f - CH_2OAc).

28-Hydroxy-*p*-friedoolean-14-en-3-one (5). Colourless needles, mp 225–227° (CHCl_3), $[\alpha]_D^{20} - 0.2^\circ$ (CHCl_3 ; c 1.00). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3350, 1080 (OH), 1708 (CO), 1640 (C=C). $^1\text{H NMR}$: δ 0.90 (3H), 0.97 (6H), 1.07 (3H), 1.08 (6H), 1.11 (3H) (all s, Me \times 7), 2.32–2.62 (2H, m, H₂-2), 3.14, 3.30 (1H each, d, $J = 10.0$ Hz, H₂-28), 5.54 (1H, dd, $J = 8.1, 3.4$ Hz, H-15). MS m/z : 440.3657 $[\text{M}]^+$ (calc. for $\text{C}_{30}\text{H}_{48}\text{O}_2$, 440.3642), 425 $[\text{M} - \text{Me}]^+$, 409 $[\text{M} - \text{CH}_2\text{OH}]^+$, 300 (ion a), 220 (ion d), 189 (base peak) (ion d - CH_2OH). Compound **5** was identified by comparison with an authentic sample derived from myricadiol (**4**) (mmp, TLC, $^1\text{H NMR}$, IR). $^{13}\text{C NMR}$: see Table I.

Acetylation of 5. A soln of **5** (5 mg) in pyridine (1 ml) was acetylated with Ac_2O (3.5 μl) at room temp. overnight. After usual work-up, the crude product was chromatographed on silica gel to give a monoacetate (**5a**) (4.0 mg), colourless needles, mp 210–211° (EtOH), $[\alpha]_D^{22} + 8.4^\circ$ (CHCl_3 ; c 0.98). (Found: C, 79.43; H, 10.69. $\text{C}_{32}\text{H}_{50}\text{O}_3$ requires: C, 79.62; H, 10.44%). IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} : 1739, 1235 (OAc), 1705 (CO). $^1\text{H NMR}$: δ 0.90 (3H), 0.96 (6H), 1.06 (3H), 1.08 (6H), 1.13 (3H) (all s, Me \times 7), 2.03 (3H, s, OAc), 2.30–2.62 (2H, m, H₂-2), 3.73 (2H, s, H₂-28), 5.47 (1H, dd, $J = 8.0, 3.0$ Hz, H-15). MS m/z : 482 $[\text{M}]^+$, 422 $[\text{M} - \text{HOAc}]^+$, 409 $[\text{M} - \text{CH}_2\text{OAc}]^+$, 300 (ion a), 262 (ion f), 202 (ion f - HOAc), 189 (base peak) (ion f - CH_2OAc). $^{13}\text{C NMR}$: see Table I.

Oxidation of 4b. A soln of **4b** (163 mg) in pyridine (20 ml) was oxidized with CrO_3 (100 mg) at room temp. overnight. After

work-up in the usual manner, the crude product was chromatographed on silica gel to give the monoacetate (**5a**, 135 mg), colourless needles, mp 210–211° (EtOH); MS m/z : 482.3767 $[\text{M}]^+$ (calc. for $\text{C}_{32}\text{H}_{50}\text{O}_3$, 482.3762).

Alkaline hydrolysis of 5a. To a soln of compound **5a** (70 mg) derived from **4b** in EtOH (15 ml) was added 1.7% NaOEt EtOH (2 ml) and the soln was allowed to stand at room temp. overnight. The solvent was removed and the residue was extracted with CHCl_3 , which was washed with 0.1 N H_2SO_4 and H_2O and then dried (Na_2SO_4). After evaporation of the solvent, the residue was chromatographed on silica gel to give compound **5** (67 mg), colourless needles, mp 224–225° (CHCl_3).

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